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PRECIOUS METAL SALTS, SOLUTIONS, ANODES AND CRYSTALS



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METALOR®

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Commercial Name	Metal	Formula	Chemical Name	CAS No.	PDS
GOLD					Au
Gold(I) ammonium sulphite solution Sulfite d'or(I) et ammonium en solution	100 g/l	$(\text{NH}_4)_3[\text{Au}(\text{SO}_3)_2] \cdot n (\text{NH}_4)_2\text{SO}_3$	ammonium gold sulphite	71662-32-3	596
Gold(III) chloride hydrate * Chlorure d'or(III) hydraté	50%	$\text{H}[\text{AuCl}_4] \cdot n \text{H}_2\text{O}$	tetrachloroauric acid	16903-35-8	594
Gold(III) chloride solution Chlorure d'or(III) en solution	250 - 300 g/l	$\text{H}[\text{AuCl}_4]$	tetrachloroauric acid	16903-35-8	599
Gold(I) cyanide * Cyanure d'or(I)	88.3%	AuCN	gold monocyanoide	506-65-0	592
Gold grains Grenailles d'or	>99.99%	Au	gold	7440-57-5	856
Gold grains Grenailles d'or	>99.999%	Au	gold	7440-57-5	866
Gold(I) potassium cyanide Cyanure d'or(I) et de potassium	68.3%	$\text{K}[\text{Au}(\text{CN})_2]$	potassium dicyanoaurate	13967-50-5	591
Gold(I) potassium cyanide solution Cyanure d'or(I) et de potassium en solution	100 g/l	$\text{K}[\text{Au}(\text{CN})_2]$	potassium dicyanoaurate	13967-50-5	593
Gold(III) potassium cyanide Cyanure d'or(III) et de potassium	57.9%	$\text{K}[\text{Au}(\text{CN})_4]$	potassium tetracyanoaurate	14263-59-3	597
Gold(III) potassium cyanide solution Cyanure d'or(III) et de potassium en solution	10/50/100 g/l	$\text{K}[\text{Au}(\text{CN})_4]$	potassium tetracyanoaurate	14263-59-3	680
Gold(III) potassium tetrachloride * Tétrachloroaurate(III) de potassium	52.12%	$\text{K}[\text{AuCl}_4]$	potassium tetrachloroaurate	13682-61-6	598
Gold(III) sodium chloride hydrate * Tétrachloroaurate(III) de sodium hydraté	49.5%	$\text{Na}[\text{AuCl}_4] \cdot n \text{H}_2\text{O}$	sodium tetrachloroaurate	15189-51-2	639
Gold(I) sodium sulphite solution Sulfite d'or(I) et sodium en solution	100 g/l	$\text{Na}_3[\text{Au}(\text{SO}_3)_2] \cdot n \text{Na}_2\text{SO}_3$	sodium gold sulphite	19153-98-1	595

SILVER					Ag
Silver(I) acetate * Acétate d'argent(I)	64.63%	AgCH_3COO	silver acetate	563-63-3	588
Silver anodes Anodes d'argent	>99.99%	Ag	silver	7440-22-4	828
Silver(I) carbonate * Carbonate d'argent(I)	78.24%	Ag_2CO_3	disilver carbonate	534-16-7	589
Silver(I) chloride Chlorure d'argent(I)	75.26%	AgCl	silver chloride	7783-90-6	584
Silver crystals 160 - 5000 μm Cristaux d'argent 160 - 5000 μm	$\geq 99.99\%$	Ag	silver	7440-22-4	684
Silver(I) cyanide Cyanure d'argent(I)	80.57%	AgCN	silver cyanide	506-64-9	582
Silver grains Grenailles d'argent	>99.99%	Ag	silver	7440-22-4	827
Silver(I) iodide * Iodure d'argent(I)	45.95%	AgI	silver iodide	7783-96-2	583
Silver(I) nitrate Nitrate d'argent(I)	63.50%	AgNO_3	silver nitrate	7761-88-8	834
Silver(I) oxide Oxyde d'argent(I)	93.10%	Ag_2O	disilver oxide	20667-12-3	587
Silver(I) potassium cyanide Cyanure d'argent(I) et de potassium	54.20%	$\text{K}[\text{Ag}(\text{CN})_2]$	potassium dicyanoargentate	506-61-6	586
Silver(I) sulphate Sulfate d'argent(I)	69.20%	Ag_2SO_4	disilver sulphate	10294-26-5	590

RUTHENIUM					Ru
Ruthenium(III) chloride hydrate * Chlorure de ruthénium(III) hydraté	38.66%	$\text{RuCl}_3 \cdot n \text{H}_2\text{O}$	ruthenium trichloride	10049-08-8	619

PLATINUM					Pt
Ammonium hexachloroplatinate(IV) Hexachloroplatinate(IV) de diammonium	43.95%	$(\text{NH}_4)_2[\text{PtCl}_6]$	diammonium hexachloroplatinate	16919-58-7	780
Hexachloroplatinic acid(IV) hydrate Acide chloroplatinique(IV) hydraté	37.8%	$\text{H}_2[\text{PtCl}_6] \cdot n \text{H}_2\text{O}$	hexachloroplatinic acid	16941-12-1	609
Hexachloroplatinic acid(IV) solution Acide chloroplatinique(IV) en solution	50/100/375 g/l	$\text{H}_2[\text{PtCl}_6]$	hexachloroplatinic acid	16941-12-1	949
Hexahydroxyplatinic(IV) acid Acide hexahydroxyplatinique(IV)	65.21%	$\text{H}_2[\text{Pt}(\text{OH})_6]$	dihydrogen hexahydroxyplatinic acid	51850-20-5	581
Platinum "P" salt 25 g/l in solution Nitrite de diammineplatine(II)	25 g/l	$[\text{Pt}(\text{NH}_3)_2(\text{NO}_2)_2]$	diammine dinitro platinum	14286-02-3	729
Potassium hexachloroplatinate(IV) Hexachloroplatinate(IV) de dipotassium	40.14%	$\text{K}_2[\text{PtCl}_6]$	dipotassium hexachloroplatinate	16921-30-5	779
Sodium hexachloroplatinate(IV) hydrate Chloroplatinate(IV) de sodium hydraté	34%	$\text{Na}_2[\text{PtCl}_6] \cdot n \text{H}_2\text{O}$	disodium hexachloroplatinate	19583-77-8	611

PALLADIUM					Pd
Palladium(II) acetate * Acétate de palladium(II)	47.41%	$\text{Pd}(\text{CH}_3\text{COO})_2$	palladium acetate	3375-31-3	683
Palladium(II) chloride Chlorure de palladium(II)	60%	PdCl_2	palladium dichloride	7647-10-1	600
Palladium(II) chloride solution Chlorure de palladium(II) en solution	60/100/200 g/l	PdCl_2	palladium dichloride	7647-10-1	839
Palladium(II) dichlorodiammine Dichlorodiammine de palladium(II)	50.34%	$[\text{Pd}(\text{NH}_3)_2]\text{Cl}_2$	diammine palladium dichloride	14323-43-3	604
Palladium(II) dichlorotetrammine Dichlorotétramine de palladium(II)	43.36%	$[\text{Pd}(\text{NH}_3)_4]\text{Cl}_2$	tetraamminepalladium dichloride	13815-17-3	605
Palladium(II) dichlorotetrammine solution Dichlorotétramine de palladium(II) en solution	100 g/l	$[\text{Pd}(\text{NH}_3)_4]\text{Cl}_2$	tetraamminepalladium dichloride	13815-17-3	608
Palladium(II) dinitrodiammine Dinitrodiammine de palladium(II)	45.77%	$[\text{Pd}(\text{NH}_3)_2(\text{NO}_2)_2]$	diammine palladium nitrite	14852-83-6	606
Palladium(II) nitrate solution Dinitrate de palladium(II) en solution	100/200 g/l	$\text{Pd}(\text{NO}_3)_2$	palladium dinitrate	10102-05-3	761
Palladium(II) oxide hydrate Monoxyde de palladium(II)	70%	$\text{PdO} \cdot n \text{H}_2\text{O}$	palladium monoxide	1314-08-5	843
Palladium(II) potassium tetrachloride Tétrachloropalladate(II) de dipotassium	32.6%	$\text{K}_2[\text{PdCl}_4]$	dipotassium tetrachloropalladate	10025-98-6	907
Palladium(II) sodium tetrachloride * Tétrachloropalladate(II) de disodium	36.17%	$\text{Na}_2[\text{PdCl}_4]$	disodium tetrachloropalladate	13820-53-6	842
Palladium(II) tetrammine sulphate Sulfatotétramine de palladium(II)	39.33%	$[\text{Pd}(\text{NH}_3)_4]\text{SO}_4$	tetramminepalladium sulphate	13601-06-4	602
Palladium(II) sulphate solution Sulfate de palladium(II) en solution	100 g/l	PdSO_4	palladium sulphate	13566-03-5	855

RHODIUM					Rh
Rhodium(III) chloride trihydrate * Chlorure de rhodium(III) trihydraté	39.08%	$\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$	rhodium trichloride	13569-65-8	618
Rhodium(III) iodide * Iodure de rhodium(III)	18.5%	RhI_3	rhodium triiodide	15492-38-3	697
Rhodium(III) sulphate in solution Sulfate de rhodium(III) en solution	50 g/l	$\text{Rh}_2(\text{SO}_4)_3$	dirhodium trisulphate	10489-46-0	914

IRIDIUM					Ir
Hexachloroiridic(IV) acid hydrate * Acide hexachloroiridique(IV) hydraté	37-39%	$\text{H}_2[\text{IrCl}_6] \cdot n \text{H}_2\text{O}$	hexachloroiridic acid	16941-92-7	693
Iridium(III) chloride hydrate * Chlorure d'iridium(III) hydraté	49-54%	$\text{IrCl}_3 \cdot n \text{H}_2\text{O}$	iridium trichloride	12645-45-3	723

* minimum order quantity

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